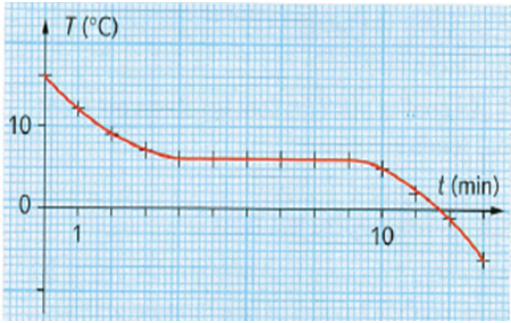
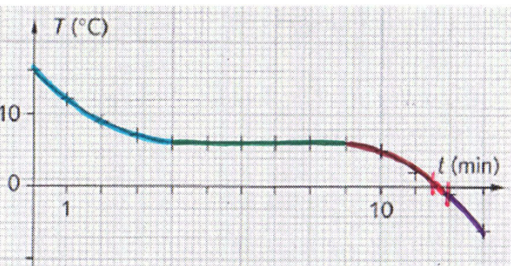
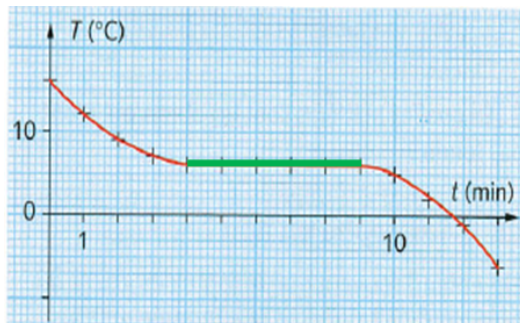


**English translation of “t-T-lesson9-quiz”
on the exploitation of a graph of the change of state**

	<p>Q1. What are the magnitudes represented on each axis? (2 answers)</p> <ol style="list-style-type: none"> 1. In ordinate (upward), the time in minutes; 2. In ordinate (upward), the temperature in Celsius degrees; 3. In ordinate (upward), the state of the matter; 4. In abscissa (rightward), the time in minutes; 5. In abscissa (rightward), the temperature in Celsius degrees; 6. In abscissa (rightward), the state of the matter; 7. Else.
	<p>Q2. How long is the experience?</p> <ol style="list-style-type: none"> 1. The experience lasts 11,5 min; 2. The experience lasts 10 min; 3. The experience lasts 16 °C; 4. The experience lasts 13 min.
	<p>Q3. How does the <i>temperature vary</i> between 16 °C and -6 °C? (2 answers)</p> <ol style="list-style-type: none"> 1. The temperature does not change (it remains the same); 2. The temperature changes; 3. The temperature increases; 4. The temperature decreases; 5. Else.
	<p>Q4. Which <i>part of the curve</i> represents the temperature plateau?</p> <ol style="list-style-type: none"> 1. The blue part; 2. The green part; 3. The red part; 4. The brown, red and purple parts; 5. Else.
	<p>Q5. Which <i>part of the curve</i> represents the change of state.</p> <ol style="list-style-type: none"> 1. The blue part; 2. The green part; 3. The red part; 4. The brown, red and purple parts; 5. Else.
	<p>Q6. Which minute does the <i>change of state</i> begin?</p> <ol style="list-style-type: none"> 1. The change of state begins at 1 min; 2. The change of state begins at 4 min; 3. The change of state begins at 6 min; 4. The change of state begins at 9 min;



5. Else.

Q7. **How long** is the *change of state*?

1. The change of state is instantaneous (it lasts 1 second);
2. The change of state lasts 5 min;
3. The change of state lasts 4 min;
4. The change of state lasts 11,5 min;
5. Else.

Q8. At what **temperature** does the state *change*?

1. The temperature of the change of state is 0 °C;
2. The temperature of the change of state is 4 °C;
3. The temperature of the change of state is 6 °C;
4. The temperature of the change of state is 11,5 °C;
5. Else.

Q9. What is the **physical state** of the substance at the beginning of the experience?

1. The achromatic liquid is hot;
2. The achromatic liquid is transparent;
3. The achromatic liquid is liquid;
4. The achromatic liquid is gaseous;
5. Else.

Q10. What is the **state** of this substance at the end of the experience?

1. The substance is cold;
2. The substance is transparent;
3. The substance is liquid;
4. The substance is solid;
5. Else.

Q11. Is this substance **water**?

1. **Yes**, this substance is water **because** the change of state of the water occurs with a temperature plateau;
2. **Yes**, this substance is water **because** the shift from the liquid state to the solid state occurs at 6 °C;
3. **Yes**, this substance is water **because** the shift from the liquid state to the solid state occurs at 0 °C;
4. **No**, this substance is not water **because** the shift from the liquid state to the solid state occurs at 0 °C ;
5. Else.