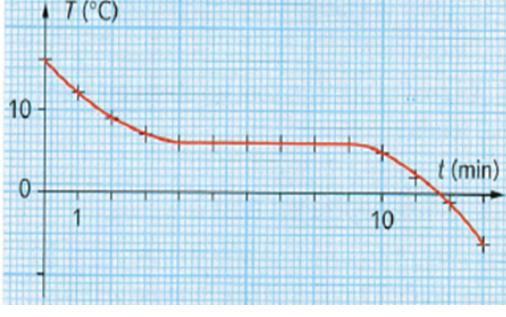
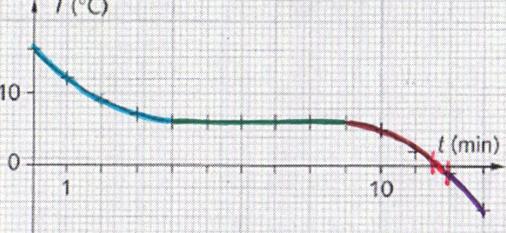
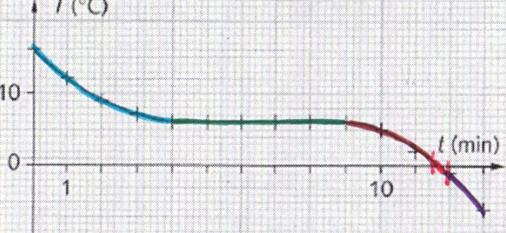
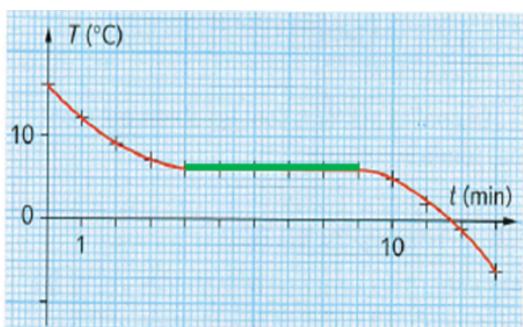


**English translation of “t-T-lesson9-quiz”**  
**on the exploitation of a graph of the change of state**

	<p>Q1. What are the <b>magnitudes</b> represented on each axis? (2 answers)</p> <ol style="list-style-type: none"> <li>1. In ordinate (upward), the time in minutes;</li> <li>2. In ordinate (upward), the temperature in Celsius degrees;</li> <li>3. In ordinate (upward), the state of the matter;</li> <li>4. In abscissa (rightward), the time in minutes;</li> <li>5. In abscissa (rightward), the temperature in Celsius degrees;</li> <li>6. In abscissa (rightward), the state of the matter;</li> <li>7. Else.</li> </ol>
	<p>Q2. <b>How long</b> is the experience?</p> <ol style="list-style-type: none"> <li>1. The experience lasts 11,5 min;</li> <li>2. The experience lasts 10 min;</li> <li>3. The experience lasts 16 °C;</li> <li>4. The experience lasts 13 min.</li> </ol>
	<p>Q3. How does the <i>temperature vary</i> between 16 °C and -6 °C? (2 answers)</p> <ol style="list-style-type: none"> <li>1. The temperature does not change (it remains the same);</li> <li>2. The temperature changes;</li> <li>3. The temperature increases;</li> <li>4. The temperature decreases;</li> <li>5. Else.</li> </ol>
	<p>Q4. Which <i>part of the curve</i> represents the <b>temperature plateau</b>?</p> <ol style="list-style-type: none"> <li>1. The blue part;</li> <li>2. The green part;</li> <li>3. The red part;</li> <li>4. The brown, red and purple parts;</li> <li>5. Else.</li> </ol>
	<p>Q5. Which <i>part of the curve</i> represents the <b>change of state</b>?</p> <ol style="list-style-type: none"> <li>1. The blue part;</li> <li>2. The green part;</li> <li>3. The red part;</li> <li>4. The brown, red and purple parts;</li> <li>5. Else.</li> </ol>
	<p>Q6. Which <b>minute</b> does the <i>change of state</i> begin?</p>
	<ol style="list-style-type: none"> <li>1. The change of state begins at 1 min;</li> <li>2. The change of state begins at 4 min;</li> <li>3. The change of state begins at 6 min;</li> <li>4. The change of state begins at 9 min;</li> </ol>



5. Else.

Q7. **How long** is the *change of state*?

1. The change of state is instantaneous (it lasts 1 second);
2. The change of state lasts 5 min;
3. The change of state lasts 4 min;
4. The change of state lasts 11,5 min;
5. Else.

Q8. At what **temperature** does the state *change*?

1. The temperature of the change of state is 0 °C;
2. The temperature of the change of state is 4 °C;
3. The temperature of the change of state is 6 °C;
4. The temperature of the change of state is 11,5 °C;
5. Else.

Q9. What is the **physical state** of the substance at the beginning of the experience?

1. The achromatic liquid is hot;
2. The achromatic liquid is transparent;
3. The achromatic liquid is liquid;
4. The achromatic liquid is gaseous;
5. Else.

Q10. What is the **state** of this substance at the end of the experience?

1. The substance is cold;
2. The substance is transparent;
3. The substance is liquid;
4. The substance is solid;
5. Else.

Q11. Is this substance **water**?

1. **Yes**, this substance is water **because** the change of state of the water occurs with a temperature plateau;
2. **Yes**, this substance is water **because** the shift from the liquid state to the solid state occurs at 6 °C;
3. **Yes**, this substance is water **because** the shift from the liquid state to the solid state occurs at 0 °C;
4. **No**, this substance is not water **because** the shift from the liquid state to the solid state occurs at 0 °C ;
5. Else.